# SAFETY DATA SHEET

**Product:** 571-P Revision Date: 6/01/2015

#### 1. MATERIAL IDENTIFICATION

Product Name: Ceramabond 571-P

**Product Description:** Off-White, Odorless Powder

Product Use: High Temperature Adhesive Powder

Manufacturer: Aremco Products, Inc.

707-B Executive Blvd. Valley Cottage, NY 10989

**Telephone:** 845-268-0039

**Emergency Phone:** 845-268-0039 or Infotrac (24/7) 800-535-5053

## 2. HAZARDS IDENTIFICATION

GHS Classification:

None

GHS Symbol:

No Pictogram Required

**GHS Signal Word:** 

None

GHS Hazard Determining Component:

None

GHS Hazard Statements:

None

**GHS Precautionary Statements - Prevention:** 

P261 Avoid breathing dust

P280 Wear protective gloves/eye protection/face protection.
P285 In case of inadequate ventilation wear respiratory protection

P302 + P352 IF ON SKIN: Wash with plenty of soap and water

P305 + P351 + P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do.

## 3. COMPOSITION

Chemical Name	CAS No.	EC No.	Concentration	GHS Product Identifier
Magnesium Oxide	1309-48-4	215-171-9	70.0-90.0%	None
Aluminum Oxide	1344-28-1	215-691-6	10.0-30.0 %	None

## 4. FIRST AID MEASURES

After eye contact: Hold eyelids open and flush with a steady, gentle stream of water for several minutes. Remove contact lenses if

present and easy to do. Continue rinsing.

After skin contact: Wash skin with soap and water.

After inhalation: In case of inhalation of dust, move to fresh air.

After ingestion: If swallowed, do not induce vomiting. If victim is conscious and alert, give 1-2 glasses of water to drink. Do not

give anything by mouth to an unconscious person. Seek medical attention immediately.

Medical Conditions Possibly Inhalation of product may aggravate existing chromic respiratory problems such as asthma, emphysema or

Aggravated by Exposure: bronchitis. Skin contact may aggravate existing skin disease.

## 5. FIRE FIGHTING MEASURES

Flash Point: Not applicable

Flammable Limits: This material is non-combustible.

Extinguishing Media: This material is compatible with all extinguishing media.

Special Fire Fighting Procedures: Protective actions and/or special protective equipment depending on surrounding fire. Use protective clothing

and self-contained breathing apparatus.

Unusual Fire & Explosion Hazards: This material is in principle not combustible, not explosive and not flammable. However, magnesium oxide may

ignite in the presence of interhalogens such as chlorine trifluoride or bromine pentafluoride. Magnesium oxide may also ignite and explode when heated with sublimed sulfur, magnesium powder or aluminum powder.

#### **6. ACCIDENTAL RELEASE MEASURES**

Personal Protection: Wear personal protective equipment to prevent inhalation of dust and skin exposure.

Spill Cleanup: Mop or sweep up spills. Place into appropriate container for disposal. Discharge in accordance with federal,

state and local regulations or permits. Avoid contact with halogens and strong acids.

## 7. HANDLING AND STORAGE

Handling: Avoid contact with eyes, skin and clothing. Avoid breathing dust. Keep container closed. Promptly clean

residue from closures with cloth dampened with water. Promptly clean up spills.

Storage: Store in a dry area in clean plastic or stainless steel containers.

### 8. EXPOSURE CONTROLS / PERSONAL PROTECTION

Chemical Name	CAS No.	EC No.	PEL (mg/m³)	TLV (mg/m³)
Magnesium Oxide	1309-48-4	215-171-9	10	15
Aluminum Oxide	1344-28-1	215-691-6	10	15

Engineering Controls: Use with adequate ventilation. Avoid dust formation. Keep containers closed. Safety shower and eyewash

fountain should be within direct access.

Respiratory Protection: If exposure limits are exceeded and local ventilation is unavailable, a supplied-air respirator or a self-contained

NIOSH-approved dust respirator is required.

Skin Protection: Wear body-covering protective clothing and gloves.

Eye Protection: Wear safety glasses with side shields.

### 9. PHYSICAL AND CHEMICAL PROPERTIES

Physical and chemical here represent typical properties of this product. Contact Technical Sales for detailed specifications.

Appearance: Powder Off-White Color: Odor: Odorless 8-10.5 pH: Specific Gravity, g/cc 3.00-3.90 Water Solubility: Insoluble Melting Point Range: Not determined **Boiling Point Range:** Not determined

**Decomposition temperature:** No available information

Auto-ignition temperature Does not ignite

Flash point: None

Flammability: Non-Flammable

## 10. STABILITY AND REACTIVITY

Chemical Stability: This material is stable under all conditions of use and storage.

Conditions to Avoid: None

Magnesium oxide may ignite in the presence of interhalogens such as chlorine trifluoride or bromine PR

phosphorous pentafluoride. Magnesium oxide may also ignite and explode when heated with sublimed sulfur, magnesium powder or aluminum powder. Exposure to strong acids may also cause vigorous reaction and heat

generation.

Hazardous Decomposition Products: Fumes may be generated if magnesium oxide is heated to the point of volatilization (> 1700 C),

Hazardous Polymerization: Will not occur.

#### 11. TOXICOLOGICAL INFORMATION

Acute Toxicity Data: Oral: None

Eye: None

Dermal: Not absorbed by intact skin. Intimate contract of naked skin to magnesium oxide

dust may cause irritation, drying and chapping.

Inhalation: Short-term inhalation of magnesium oxide dust or fume may cause temporary

irritation of upper respiratory track, skin, nose and eyes. No known allergic

responses.

Chronic Toxicity Data: Oral: Magnesium oxide: After repeated exposure, man, 800 mg/m3, no observed

effect. After repeated exposure, cattle, target organ: gastro-intestinal system,

1% irritating effect.

Inhalation: Magnesium oxide: After repeated exposure, rat, 3 mg/m3, no observed effect.

Eye: May cause eye irritation.

Mutagenic Effects: No known studies. Not considered to be mutagenic in general.

Carcinogenic Effects: Substance is not classified as carcinogenic under ACGIH, NIOSH, IARC, NTP or

OSHA.

#### 12. ECOLOGICAL INFORMATION

Ecotoxity: Not water endangering. Aquatic toxicity is unlikely due to low solubility.

Persistence & Degradability: Magnesium oxide reacts with water to produce magnesium hydroxide. The reaction is self-limiting because of

the formation of insoluble magnesium hydroxide. No other data concerning degradation are available.

Bioaccumulative Potential: Not expected.

Mobility in Soil: Not expected.

**PBT and vPvB Assessment:** No information available.

#### 13. DISPOSAL CONSIDERATIONS

Disposal Method: Dispose in accordance with federal, state and local regulations and permits.

Evaporate water from sol and dispose of the solids in a landfill.

## 14. TRANSPORTATION INFORMATION

**DOT UN Status:** This material is not a regulated hazardous material for transportation.

## 15. REGULATORY INFORMATION

#### **U.S. Federal Regulations**

CERCLA: No CERCLA reportable quantity has been established for this material.

TSCA: All ingredients of this material are listed on the TSCA inventory.

SARA Title III

Sections 302, 304, 313: This product does not contain any substances reportable under these sections.

Sections 311, 312:

Hazard Classes	Yes/No
Fire Hazard	No
Reactivity Hazard	No
Pressure Hazard	No
Immediate Hazard	No
Delayed Hazard	No

International Inventory	<u>Status</u>
Canada (DSL)	Yes
Europe (EINECS/ELINCS)	Yes
Australia (AICS)	Yes
Japan (MITI)	Yes
South Korea (KECL)	Yes

#### **16. OTHER INFORMATION**

NFPA Ratings (Scale 0–4)

Health, 1

Flammability, 0 Reactivity, 0

Personal Protection, F

HMIS Ratings (Scale 0–4) Health, 1

Flammability, 0 Reactivity, 0

Personal Protection, F





#### **Key Legend Information**

ACGIH American Conference of Governmental Industrial Hygienists

ARD International Agency for Research on Cancer

CAS Chemical Abstract Service

CERCLA Comprehensive Environmental Response, Compensation & Liability Act

DSL Domestic Substance List EC European Commission

HMIS Hazardous Materials Identification System

IARC International Agency for Research on Cancer

ND Not Determined NE Not Established

NFPA National Fire Protection Association

NIOSH National Institute for Occupational Safety & Health

NTP National Toxicology Program

OSHA Occupational Safety and Health Administration

PEL Permissible Exposure Limit

RE Repeat Exposure

RTECS Registry of Toxic Effects of Chemical Substances
SARA Superfund Amendments & Reauthorization Act
SARA Title III Emergency Planning & Community Right to Know Act

SARA Section 302 Extremely Hazardous Substances

SARA Section 304 Emergency Release

SARA Section 311 MSDS/List of Chemicals & Hazardous Inventory

SARA Section 312 Emergency & Hazardous Inventory
SARA Section 313 Toxic Chemicals & Release Reporting

SE Single Exposure

STEL Short Term Exposure Limit
STOT Specific Target Organ Toxicity
TLV Threshold Limit Value
TWA Time Weighted Average

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